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Key indicators

Single-crystal X-ray study
 $T = 296$ K
Mean $\sigma(\text{C}-\text{C}) = 0.006$ Å
 R factor = 0.054
 wR factor = 0.123
Data-to-parameter ratio = 8.0

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

(*E*)-2-Methoxy-6-[(2-trifluoromethylphenyl)imino)methyl]phenol

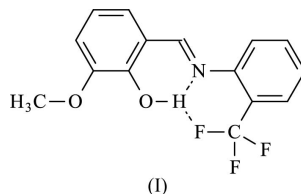
The title compound, $\text{C}_{15}\text{H}_{12}\text{NO}_2\text{F}_3$, adopts the phenol–imine tautomeric form. The two aromatic rings are twisted with respect to each other, with a dihedral angle of $17.83(19)^\circ$. $\text{O}-\text{H}\cdots\text{N}$ and $\text{O}-\text{H}\cdots\text{F}$ intramolecular interactions are found in the molecule, and the solid-state structure is stabilized by intermolecular $\text{C}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\text{F}$ hydrogen bonds, to form a two-dimensional network.

Comment

Much work has been devoted to the physicochemical characterization of substituted aromatic Schiff bases, because these compounds show remarkable photochromic properties, mainly in the solid state (Cohen *et al.*, 1964; Hadjoudis *et al.*, 1987). Photochromism arises from intramolecular H-atom transfer, together with a change in the π -electron configuration. The effect of intermolecular interactions, such as π - π charge transfer or hydrogen bonding, on H-atom transfer processes has been investigated in the solid state (Hadjoudis *et al.*, 1987; Puranik *et al.*, 1992). *N*-substituted *o*-hydroxyimines have been reported to display photochromism and thermochromism in the solid state by H-atom transfer from the hydroxy O atom to the N atom (Hadjoudis *et al.*, 1987; Xu *et al.*, 1994).

Numerous trifluoromethyl aromatic compounds have higher melting points, sometimes significantly higher, than the corresponding methyl-substituted compounds. Despite their lower molecular weight, trifluoromethyl compounds generally also have higher boiling points. Apparently, stronger forces exist in the crystalline state between molecules containing a CF_3 group than in the methyl analogues. These forces do not extend to the liquid state (Boitsov *et al.*, 2002).

o-Hydroxyarylidene Schiff bases demonstrate two possible tautomeric forms, keto–amine and phenol–imine. In the solid state, the keto–amine tautomer has been established in naphthaldimine compounds (Hökelek *et al.*, 2000; Ünver *et al.*, 2002) while the phenol–imine tautomer is found in salicylaldimine Schiff bases (Dey *et al.*, 2001; Yang & Vittal, 2003; Karadayı *et al.*, 2003). Against this background, we present here the crystal structure of the title compound, (I).



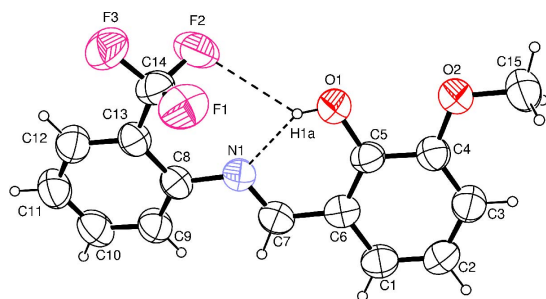


Figure 1

A view of the molecule of (I), showing the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level. Dashed lines indicate hydrogen bonds.

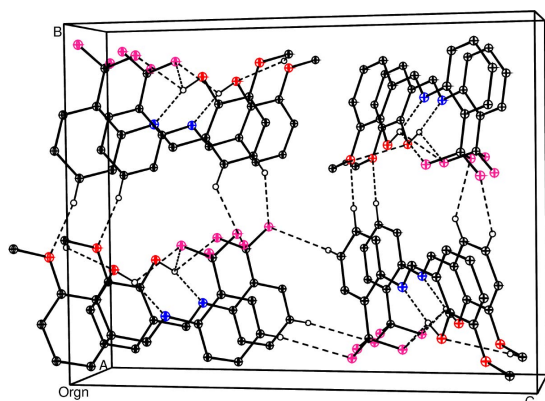


Figure 2

A packing diagram for (I). Dashed lines indicate hydrogen bonds. H atoms not involved in hydrogen bonding have been omitted for clarity.

view of the molecule of (I) and a packing diagram are shown in Figs. 1 and 2, respectively. The C8–N1 and C7–C6 bond lengths are 1.418 (5) and 1.454 (5) Å, respectively (Table 1), and agree with the corresponding distances in 2-[(4-hydroxyphenyl)iminomethyl]thiophene [1.422 (2) and 1.447 (3) Å, respectively; Kazak *et al.*, 2000], *trans*-4-[(4-dimethylamino-phenyl)iminomethyl]-*N*-methylpyridinium *para*-toluenesulfonate [1.413 (3) and 1.461 Å, respectively; Coe *et al.*, 2001], *N*-[3,5-bis(trifluoromethyl)phenyl]salicylaldimine [1.417 (4) and 1.439 (4) Å, respectively; Karadayı *et al.*, 2003] and *N*-[3,5-bis(trifluoromethyl)phenyl]-3-methoxysalicylaldimine [1.412 (4) and 1.442 (4) Å, respectively; Karadayı *et al.*, 2003]. The N1=C7 bond length of 1.270 (5) Å is typical of a double bond, similar to the corresponding bond lengths in *N*-(2,4-dinitrophenyl)-*N*-methylhydrazine [1.279 (2) Å; Aygün *et al.*, 1998], 2-salicylideneamino-4,5,6,7-tetrahydrobenzo[*b*]thiophene-3-carbonitrile [1.281 (2) Å; Elerman & Elmali, 1998] and *N*-[3,5-bis(trifluoromethyl)phenyl]salicylaldimine [1.276 (4) Å; Karadayı *et al.*, 2003]. The O1–C5 and O2–C4 bond lengths are 1.346 (4) and 1.367 (4) Å, respectively, and agree with the corresponding distances in 2-[[tris(hydroxymethyl)methyl]aminomethylene]cyclohexa-3,5-dien-1(2*H*)-one and its 6-hydroxy and 6-methoxy derivatives [1.3025 (16) and 1.3684 (19) Å, respectively; Odabaşoğlu *et al.*, 2003]. The O2–C4 and O2–C15 bond lengths are 1.367 (4) and 1.424 (5) Å, respectively, and these are similar to the corresponding distances in (*E*)-2-hydroxy-3,3',4'-trimethoxy-

stilbene [1.371 (2) and 1.427 (3) Å, respectively; Stomberg *et al.*, 1998]. The dihedral angle between the C1–C6 and C8–C13 benzene rings is 17.83 (19)°.

Compound (I) forms two strong intramolecular hydrogen bonds [O···N 2.625 (4) Å and O···F 3.199 (4) Å; Table 2], with C5–C6–C7–N1 and C7–C6–C5–O1 torsion angles of –0.6 (5)° and 0.8 (6)°, respectively (Table 1). There are also intermolecular C···F and C···O hydrogen bonds present, producing a two-dimensional network (Fig. 2, Table 2).

Experimental

The title compound, (I), was prepared by refluxing a solution prepared from *o*-vanillin (0.5 g, 3.286 mmol) in ethanol (10 ml) and 2-(trifluoromethyl)aniline (0.529 g, 3.286 mmol) in ethanol (10 ml). The reaction mixture was stirred for 1 h under reflux and left to cool. The product, (I), was recrystallized from methanol (yield 73%; m.p. 390–392 K).

Crystal data

C₁₅H₁₂F₃NO₂
M_r = 295.26
 Orthorhombic, *P*2₁2₁2₁
a = 5.2070 (5) Å
b = 14.0918 (18) Å
c = 18.1737 (18) Å
V = 1333.5 (3) Å³
Z = 4
D_x = 1.471 Mg m^{–3}

Mo *K*α radiation
 Cell parameters from 3996 reflections
 θ = 1.8–25.5°
 μ = 0.13 mm^{–1}
T = 296 K
 Block, orange
 0.50 × 0.29 × 0.18 mm

Data collection

Stoe IPDS 2 diffractometer
 φ scans
 Absorption correction: integration
 (*X-RED32*; Stoe & Cie, 2002)
T_{min} = 0.954, *T_{max}* = 0.979
 8826 measured reflections
 1562 independent reflections

931 reflections with *I* > 2σ(*I*)
R_{int} = 0.142
 θ_{\max} = 26.0°
h = –6 → 6
k = –17 → 16
l = –22 → 22

Refinement

Refinement on *F*²
R [*F*² > 2σ(*F*²)] = 0.054
wR (*F*²) = 0.123
S = 0.89
 1562 reflections
 196 parameters
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0641P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$
 (Δ/σ)_{max} < 0.001
 $\Delta\rho_{\max} = 0.22 \text{ e \AA}^{-3}$
 $\Delta\rho_{\min} = -0.23 \text{ e \AA}^{-3}$
 Extinction correction: *SHELXL97*
 (Sheldrick, 1997)
 Extinction coefficient: 0.015 (3)

Table 1

Selected geometric parameters (Å, °).

C1–C6	1.413 (5)	C6–C7	1.454 (5)
C3–C4	1.370 (5)	C7–N1	1.270 (5)
C4–O2	1.367 (4)	C8–C13	1.414 (5)
C4–C5	1.409 (5)	C8–N1	1.418 (5)
C5–O1	1.346 (4)	C12–C13	1.392 (6)
C5–C6	1.378 (5)	C15–O2	1.424 (5)
O2–C4–C5	114.9 (3)	C13–C8–N1	116.7 (3)
O1–C5–C6	123.1 (4)	C12–C13–C14	119.9 (4)
C5–C6–C7	121.6 (4)	C8–C13–C14	121.2 (4)
C1–C6–C7	118.4 (3)	C7–N1–C8	120.5 (3)
N1–C7–C6	122.0 (3)	C4–O2–C15	116.9 (3)
C9–C8–N1	124.5 (4)		
O1–C5–C6–C7	0.8 (6)	C5–C6–C7–N1	–0.6 (5)

Table 2
Hydrogen-bond geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
O1—H1A...N1	1.00 (5)	1.69 (5)	2.625 (4)	154 (4)
O1—H1A...F2	1.00 (5)	2.57 (5)	3.199 (4)	121 (3)
C2—H2...F3 ⁱ	0.93	2.79	3.650 (5)	154
C1—H1...F1 ⁱ	0.93	2.77	3.443 (4)	130
C10—H10...O2 ⁱⁱ	0.93	2.47	3.342 (5)	157
C11—H11...F3 ⁱⁱⁱ	0.93	2.90	3.751 (5)	153
C15—H15B...O1 ^{iv}	0.96	2.88	3.724 (6)	148

Symmetry codes: (i) $-x + 2, y - \frac{1}{2}, -z + \frac{1}{2}$; (ii) $-x + 1, y - \frac{1}{2}, -z + \frac{1}{2}$; (iii) $x - \frac{1}{2}, -y + \frac{3}{2}, -z$; (iv) $x + 1, y, z$.

The absolute configuration was not determined from the X-ray data, as no strong anomalous scatterer is present; 1069 Friedel pairs were merged before the final refinement. Atom H1A bound to O1 was refined freely. All H atoms bound to C atoms were refined using a riding model, with C—H = 0.93 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for aromatic C, and C—H = 0.96 Å and $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$ for methyl C atoms.

Data collection: *X-AREA* (Stoe & Cie, 2002); cell refinement: *X-AREA*; data reduction: *X-RED32* (Stoe & Cie, 2002); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP3* for Windows (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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